Physiological and histological effects of (zinc and iron) oxide nanoparticles on some fertility parameters in female mice

Shaima R.Ibraheem¹, Muntaha R. Ibrahim²

¹Biotechnology department, Collage of Science, Baghdad University, IRAQ ²Biomedical department, Alkawarizmi Collage of engineering, Baghdad University, IRAQ

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ABSTRACT

Nowadays nanoparticles have widespread application in various industries because of their special and unique features, there are many studies in side effects of nanomaterial. This study done by 40 white female mice with every other day intraperitoneally injection of low and high doses of both of ZnO nanoparticles (20 and 150 μ g/kg of body weight) and FeO nanoparticles (5 and 40 mg/kg). After a 15 days period, the mice were sacrificed and blood samples were collected for hormone analysis, and tissue samples for morphometric studies.

Statistical Analysis shows significant differences in LH, Estrogen, Progesterone hormone levels between groups, while there are insignificant differences in Follicle stimulating hormone (FSH) level between the groups compared with its level in the control group.

The results also show that the highest level of LH reach 7.2 mIU/ml in the groups treated with low dose of zinc oxide, the highest level of FSH reach 4.58 mIU/ml in the groups treated with low dose of zinc oxide, the highest level of Estrogen hormone reach 69.5 ng/ml in the groups treated with low of dose zinc oxide and the highest level of Progesterone reach 1.9 ng/ml in the groups treated with high dose iron oxide. We conclude from the results that the low doses of ZnO has benefits in increasing fertility through high level of reproductive hormones, while the high levels of nanoparticles reduce fertility and there is a relation between FeO nanoparticles and progesterone levels which may need more future studies.

Morphometric study of the ovary show increase in Follicular stages number range in the group treated with Low dose ZnO in compare with its range in the control groups. The lower range was belong to the group treated with the high dose of FeO. No significant differences has been found in the diameter mean of the different follicular phases between the group treated with low dose of ZnO NPs in compared with the control group. High dose of ZnO NPs cause significant increase in the diameter mean of Primordial follicles in compared with the control group. Low and high dose FeO NPs treated groups show significant reduction in the diameter mean of the different follicular phases in compared with the control group.

Keywords: Nanoparticles, fertility, hormones, mice.

الخلاصة

للجزيئات النانوية في الوقت الحالي تطبيقات متنوعة وفي مختلف الصناعات وذلك بسبب خصائصها الفريدة وقد استحدثت العديد من الدراسات التي تهتم بكشف التاثيرات الجانبية والتاثيرات الفسيولوجية لهذه الجزيئات. شملت الدراسة الحالية 40 من اناث الفئران البيض تم حقنها يوميا بمقدار 0.1 ملليلتر في البريتون بجرعات قليلة وعالية من جزيئات اوكسيد الزنك النانوية (بتركيز 20 و 150 ميكروغرام /كيلوغرام من وزن الجسم) و جزيئات اوكسيد الحقن تم النانوية (بتركيز 5 و 40 ميكروغرام /كيلوغرام من وزن الجسم) و بعد 15 يوم من الحقن تم دراسات مول على عينات الدم من الفئران لغرض اجراء الاختبارات الهرمونية وعينات نسيجية لاجراء دراسات مورفوميترية.

اظهرت الدراست الاحصائية وجود فروقات معنوية في مستويات هرمونات (هرمون الاباضة LH و الاستروجين E2 والبروجسترون Prg) مقارنة مع مستوياتها في مجموعة السيطرة في حين لم توجد اي فروقات معنوية في مستوى الهرمون المحفز للجريبات FSH بالمقارنة مع السيطرة . وجد من خلال نتائج البحث ان اعلى مستوى لهرمون المحفز للجريبات FSH بالمقارنة مع السيطرة . وجد من خلال نتائج البحث ان اعلى مستوى لهرمون الاباضة بلغ FSH 7.2 mIU/ml في محفز الجريبات http:// المحفز الجريبات FSH تعاقب معنوية في مستوى المحاميع وجد من خلال نتائج البحث ان اعلى مستوى لهرمون الاباضة بلغ FSH 7.2 mIU/ml في المحاميع المعاملة بجرع واطئة لجزيئات اوكسيد الزنك النانوية و اعلى مستوى لهرمون محفز الجريبات http:// المحاميع المعاملة بجرع واطئة لجزيئات اوكسيد الزنك النانوية و اعلى مستوى لهرمون محفز الجريبات واعلى مستوى لهرمون محفز الجريبات واعلى مستوى لهرمون الاباضة بلغ معاملة بجرعة واطئة لجزيئات اوكسيد الزنك النانوية و اعلى مستوى لهرمون محفز الجريبات المعامملة بجرع واطئة لجزيئات اوكسيد الزنك النانوية و اعلى مستوى لهرمون الاستروجين بلغ المحموعة المعاملة بجرعة واطئة ليزيئات اوكسيد الزنك النانوية و اعلى مستوى لهرمون الاستروجين بلغ المحموعة المعاملة بجرعة واطئة لمتارينات اوكسيد الزنك النانوية المعنوي لهرمون البرجستيرون بلغ المعاملة بجرعة واطئة الجزيئات اوكسيد الزيئات اوكسيد الرون المعموعة المعاملة بجرعة واطئة المرويات المعامية بجرعة عالية لجزيئات اوكسيد الحديد النانوية. تشير هذه النتائج الى كون المحموعة المعاملة بجرعة عالية لجزيئات اوكسيد الحديد النانوية المعاملة بجرعة ما معن ملال رفع مستويات هرمونات الخصوبة ،بينما الجرعات العالية تسبب اختز ال مستويات هرمونات. ومعنوي الخصوبة من خلال رفع مستويات هرمونات الخصوبة ،بينما الجرعات العالية تسبب اختز ال مستويات ها مموزي رفع مي المرمونات. وكسيد الزنك النانوية معاني ما معنويات ما معنويات وكسيد الزنك النانوية لها فائدة في زيادة معايير الخصوبة من خلال رفع مستويات هرمونات الخصوبة ،بينما الجرعات العالية تسبب اختز ال مستويات هذه الهرمونات. وهي حقيقة تحتاج لمزيد من الدراسة في المستقبل.

وجد من خُلال دراسة المقاطع النسيجية لمبايض الفَنران زيادة في اعداد المرحلة الجريبية لدى المجاميع المعاملة بجرعات واطنة من جزيئات اوكسيد الزنك النانوية وبالمقارنة مع اعدادها لدى مجموعة السيطرة واقل اعداد لهزه المرحلة وجدت لدى المجموعة المعاملة بجزيئات اوكسيد الحديد النانوية وبالمقارنة مع اعدادها لدى مجموعة السيطرة. لم تلاحظ فروقات معنوية في معدل قطر مختلف مراحل البيضة لدى المجموعة المعاملة بجرعة واطنة من جزيئات اوكسيد الزنك النانوية وبالمقارنة مع معموعة المعاملة بجرعة واطنة من جزيئات اوكسيد الزنك النانوية وبالمقارنة مع مجموعة السيطرة، في حين الجرعة العالية من جزيئات اوكسيد الزنك النانوية مسببت زيادة معنوية في معدل قطر المرحلة الاولية للبيضة وبالمقارنة مع مجموعة السيطرة. أما الجرعات الواطنة والعالية لجزيئات اوكسيد الحديد النانوية سببت اختزالا معنويا في معدل قطر مختلف مراحل البيضة وبالمقارنة مع مجموعة الميطرة.

INTRODUCTION

Small size and high surface area of Nanoparticles make it a principle participant in all aspects of modern life application [1]. Because of size dependent chemical and properties, nanoparticles physical are considered an interesting candidates for applications in both in vivo and in vitro biomedical research. In the field of medicine, nanoparticles implicated mainly in targeted drug delivery, imaging, sensing, and artificial implants, use as antimicrobials to target highly pathogenic and drug resistant microbes [2]. The application of nanoparticles on health and environment needs to be assessed before their large-scale production and application in various fields [3-4].

Nanoparticles rapidly can enter the bloodstream and reach to the organs (including the brain, heart, kidneys) [5] Experiment some of injected nanoparticles show that they have negligible side effect [6]. High surface area and other features of nanoparticles cause it is very reactive and toxic and can damage human and animal cells by increasing oxidative stress mechanism [1].

Among the various metal oxides studied for their medicinal application, zinc oxide nanoparticles have been found to be highly toxic [7-8]. ZnO NPs are used in many commercial products such as cosmetics, paints, textiles, and sanitary products. ZnO NPs are used in nutrients as additives. They have antifungal property which are used as pesticides in agriculture sectors and is also deployed in anticancer characteristics in medicines [9].

with other nanoparticles Iron many characterized by its magnetic properties and stability [10]. Iron oxide nanoparticles have widespread application for in vivo and in vitro research due to the physiochemical characteristics [11], and used for drug delivery in cancer therapy since 1970 [12-13]. Yet, there is a few studies have been done about the effects of nanoparticles on the reproductivity potential [14]. The goal of this study is to evaluate the effect of low and high doses of ZnO and FeO nanoparticles on fertility in female mice.

MATERIALS AND METHODS Chemicals and kits:

- Zinc oxide nanoparticles (ZnO NPs) solution (Conc. Wt % in water, size 35 nm) was obtained from Sigma, USA.
- Iron oxide nanoparticles (Fe₂O₃ NPs) solution (Conc. 20 wt% in water, size 30 nm) was obtained from Sigma, USA.

Solutions preparations:

Nanoparticles solutions: The solutions of nanoparticles were prepared by distilled water with a minimum concentration ($20 \ \mu g/kg$) of ZnO for low dose group and with a maximum concentration ($150 \ \mu g/kg$) of ZnO for high dose group. FeO nanoparticles solutions prepared by the same way with distilled water (low dose concentration 5 mg/kg and high dose concentration 40 mg/kg) [15-16].

Phosphate buffer saline (PBS): (8g of NaCl, 0.2g of KCl, 1.44g of Na2HPO4, 0.24g of KH2PO4) .Dissolve the following in 1L distilled H2O ,Adjust pH to 7.4 with HCl.

Experiment design: Fifty adult female mice were obtained from the animal house of National center for drug control and research and kept in the animal house of biotechnology Research Center, Al-Nahrain University ,where its placed in a separated cages at $(25C^{\circ})$ room temperature, and fed with suiTable diet in addition to water.

The experimental animals were randomly divided into five groups as follows:

- First group: Control group consisted of 10 animals were injected intraperitonially with PBS for 15 days.
- Second group: consisted of 10 animals were injected intraperitonially with 0.1 ml of ZnO nanoparticles solution (20 µg/kg) for 15 days., represent the low dose group.
- Third group: consisted of 10 animals were injected intraperitonially with 0.1 ml of ZnO nanoparticles solution (150 μ g/kg) for 15 days., represent the high dose group.
- Fourth group: consisted of 10 animals were injected intraperitonially with 0.1 ml of FeO nanoparticles solution (5 mg/kg) for 15 days., represent the low dose group.
- Fifth group: consisted of 10 animals were injected intraperitonially with 0.1 ml of FeO nanoparticles solution (40 mg/kg) for 15 days., represent the high dose group.

Specimen collection and processing:

Blood samples were collected for hormonal analysis. Serum was separated bv centrifugation at 3000 rpm for 15 min and stored at - 20° C until analysis. Luteinizing hormones (LH), Follicle stimulating hormones (FSH), Estrogen, Progesterone hormones levels were estimated using ELIZA kit (Orgmetric Germany) in accordance with the manufacturer's recommendations.

The ovaries of three mice of each group were immediately excised and preserved in 10% formalin for morphometric study which is conducted according to the method used by [17]. Eight microns thick sections were cut and stained with hematoxylin and eosin for microscopic examination.

Morphometric study were achieves using ocular micrometer

Ovaries of three animals of each group besides the control were taken for the follicular studies. Sections of the ovary were examined under a light microscope and the general histologic appearance of the ovary was assessed. All sections of the ovary were counted for various stages of development of follicles as described by Junqueira *et al..*, 1992 [18]. Four classes of ovarian follicles were categorized as follows:

• Primordial follicle (PF): contained an oocyte surrounded by a partial or complete layer of squamous granulosa cells.

- Primary follicle: showed a single layer of cuboidal granulose cells. "Growing" follicles had an oocyte surrounded by a multilayered, solid mantle of granulosa cells.
- Secondary follicles were surrounded by more than one layer of cuboidal granulosa cells and accumulations of follicular fluid appear between the granulosa cells.
- Antral follicles were characterized by a central oocyte enriched by a fluid-filled space and bordered by hundreds of layered granulosa cells.

By using these criteria, mean diameter of each follicle stage have been measured by using a calibrated ocular micrometer to avoid repeated counting. (see Figure 1).



Figure 1: Histological section of mice ovary $(20\times)$ represents different stages of follicles. A:Primordial, B: Primary follicle C: Secondary follicle, D: Antral follicle.

Statistical Analysis: The Statistical Analysis System-SAS (2010) was used to study the effect of different factors in studied parameters. Least significant difference –LSD test was used to significant compare between means in this study. Results were expressed as mean \pm SE. Values were considered to be statistically significant at P<0.05 [19].

RESULTS AND DISSCUSSION: Reproductive hormones analysis:

Statistical Analysis of our results (see Table 1) shows significant differences in LH and estrogen levels between the groups and its levels in the control group.

The results also show that the highest level of LH reach 7.20 ± 0.34 mIU/ml in the groups treated with low dose zinc oxide, the highest level of FSH reach 4.58 ± 0.30 mIU/ml in the groups treated with low dose zinc oxide, the

highest level of Estrogen hormone reach 69.50 ± 2.10 ng/ml in the groups treated with low dose zinc oxide and the highest level of Progesterone reach 1.90 ± 0.08 ng/ml in the groups treated with high dose iron oxide.

Estrogen hormone reaches its lowest level in the group treated with high dose of ZnO. There are no significant differences noticed in the FSH and Progesterone levels between the groups and in compare with its levels in the control group.

These results agree with the results of (Esmaeillou *et al.*, 2013) [20]. In this research, there is no alteration in serum sex hormone levels such as FSH, LH and estradiol in female Wistar rats after administration with 333.33 mg/kg zinc oxide NPs (ZnO-NPs).

Our study showed that low doses of ZnO cause nanoparticles increase in the reproductivity through the high levels of LH, FSH and estrogen, while the high dose causes low level of these hormones, this referred to decreased in reproductivity as mentioned by Espanani et al., 2014 [21]. In his research, he confirmed that exposing to ZnO nanoparticles damaged to public health and reduced fertility potential. Puran et al., 2013 [22] referred to the protective effect of zinc oxide nanoparticles through its antioxidant potential against the adverse effects of the anticancer drug, which is known as doxorubicin (DOX), one of its adverse effects is male infertility [23]. Administration of ZnO significantly improved DOX-induced changes in plasma total antioxidant power (TAP), Lipid peroxidation (LPO), plasma testosterone, LH, sperm count and DNA damage.

The zinc is one of the seven main trace elements besides to copper, iron, cobalt, iodine, manganese and selenium among these, iron (1.0-2.0 ppm) is the most abundant in serum followed by zinc (0.8-1.2 ppm) [24].

However, toxicological studies have shown that ZnO nanoparticles could be harmful to human and other species [25]. Han *et al..*, 2011 [26] reported neurotoxicity in rats after intraperitoneal administration of ZnO nanoparticles. Many studies have reported the cytotoxic and genotoxic effects of ZnO nanoparticles in various mammalian cell lines [27-28]. Results showed the benefits of using nanoparticles in low doses which may have a protective effect through the antioxidant mechanisms, at the same time the high doses may have a harmful effects on many levels. This definitely requires more studies for the mechanisms that may cause this harm.

Zinc has an essential role for thyroid hormone secretion and function and other hormones such as testosterone, insulin, and growth hormone [29-30]. Also, zinc plays an essential role in sexual development. Copper and zinc important role in regulating play an progesterone production by luteal cells via involvement of superoxide dismutase [31]. Zinc is involved in there organization of ovarian follicles which are the source of progesterone. This occurs through the involvement of metalloproteinase-2 (MMP-2) enzyme, which is a member of zinc endopeptidase family [32].

Table 1: Hormones Concentrations between the different groups.

	Mean ± SE			
Group	LH (mIU/ ml)	FSH (mIU/ ml)	Estroge n (ng/ml)	Progestero ne (ng/ml)
Control	6.70 ± 0.79 ab	4.45 ± 0.37 a	69.00 ± 2.48 a	1.52 ± 0.28 a
Low dose ZnO	7.20 ± 0.34 a	4.57 ± 0.30 a	69.50 ± 2.10 a	1.75 ± 0.22 a
High dose ZnO	6.67 ± 0.54 ab	4.32 ± 0.39 a	35.00 ± 2.34 b	1.57 ± 0.21 a
Low dose FeO	6.35 ± 0.46 ab	4.55 ± 0.29 a	68.00 ± 2.67 a	1.60 ± 0.28 a
High dose FeO	5.60 ± 0.25 b	4.16 ± 0.24 a	67.66 ± 3.39 a	1.90 ± 0.08 a
LSD value	1.033 *	0.882 NS	16.752 *	0.579 NS
* (P<0.05), NS: Non-significant.				
*Significant NS : Not Significant				

After short-terms of exposure to zinc, histological alterations were observed in organs like the pancreas, pituitary gland, and adrenal gland of zinc-exposed mice fed with zinc sulphate diet [33]. Straube *et al.*. (1980) [34]

also examined the effects of excess dietary zinc in adult ferrets fed a diet of canned dog food (that contained 27 ppm zinc and 3.3 ppm copper). Animals in high doses between (1500 and 3000) ppm showed signs of severe toxicity represented by (macrocytic hypo-chromic anemia, increased reticulocyte count, diffuse nephrosis, and the presence of protein, glucose, blood, and bilirubin in the urine, increases in tissue zinc levels, decreases in copper levels, and decreased ceruloplasmin oxidase activity). Increasing use of iron oxide nanoparticles in medicine and environmental remediation has led to concerns regarding exposure of these nanoparticles to the public [35]. Besides the magnetic properties of iron (II, III) oxide nanoparticles (IONPs) it is also have an abundant potential in several biomedical and clinical applications, such as magnetic resonance imaging (MRI), tissue repair, as drug delivery, and magnetic hyperthermia cancer treatment [36].

However, possible risk on fertility hormones due to increasing application of Fe_2O_3 is not fully elucidated. Table 1 shows that no significant differences have been found between hormones levels in the low dose of FeO NPs treated group and control group. While high dose cause significant decrease in LH level in comparison with the control.

Recent studies reported that FeO NPs have the ability to induce genotoxicity which is likely to be mediated through ROS generation and oxidative stress in a dose-dependent manner, evident by depletion of glutathione lipid peroxidation [37]. However, in less than seven days the excess iron was either excreted or incorporated into the body [38]. (In the form of co-factors, hemoglobin, *etc.*).

Changes of oxidative stress were observed to be dependent on the body mass index (BMI) and the duration of infertility of the enrolled women. Levels of malondialdehyde (MDA) correlate negatively with LH concentration; meaning the increase in oxidative stress followed by regression in LH levels, suggesting that oxidative stress is involved in the pathophysiology of infertility in females, particularly through the induction of changes in gonadotrophin hormones [39].

Morphometric study of the ovary

Table 2 shows increase in Follicular stages number range in the group treated with Low dose ZnO in compare with its range in the control groups. Other groups' show a decrease in the Follicular stages number range and the lower range was belong to the group treated with the high dose of FeO, this may refer to the damaged caused by FeO precipitation in ovary tissue.

No significant differences has been found in the diameter mean of the different follicular phases between the group treated with low dose of ZnO NPs in compared with the control group, for example: (The diameter of primodial phase is 92.50 μ m \pm 11.08 in Low dose ZnO group in comparison to control group 95.00 µm \pm 20.20). High dose of ZnO NPs cause significant increase in the diameter mean of Primordial follicles (160.00 $\mu m \pm 9.12$) in compared with the control group (95.00 μ m \pm 20.20). Low and high dose FeO NPs treated groups show significant reduction in the diameter mean of the different follicular phases in compared with the control group as show in Table 3, for example: (The diameter of Antral follicle phase is 595.00 μ m \pm 25.98 in low dose FeO group in comparison to control group 710.00 μ m ± 102.14).

 Table 2: Number ranges of follicular stages in mice

 ovary in the different treated groups

	Groups					
Follicl e stage	G1/Co ntrol	G2/ Low dose ZnO	G3/ High dose ZnO	G4/ Low dose FeO	G5/ High dose FeO	
Primo rdial	9-18	11-20	5-7	5-7	1-2	
Prima ry follicle	8-16	10-18	6-7	6-7	1-2	
Secon dary follicle	6-8	10-12	4-5	4-5	2-3	
Antral follicle	1-3	2-3	1-2	1-2	0	

Table 3: Follicles diameter (Micrometer) of different stages of mice ovary in the different groups

	Mean (μm)± SE			
Group	Primordia l	Primar y follicle	Secondar y follicle	Antra l follicl e

Contro l	95.00 ± 20.20 b	290.00 ± 32.14 a	366.67 ± 32.99 a	710.00 ± 102.14 a	
Low dose ZnO	92.50 ± 11.08 b	286.67 ± 18.55 a	343.33 ± 23.92 a	736.67 ± 86.85 a	
High dose ZnO	160.00 ± 9.12 a	288.00 ± 15.27 a	367.50 ± 13.67 a	693.33 ± 82.93 a	
Low dose FeO	80.00 ± 9.12 b	166.67 ± 24.03 c	345.00 ± 45.55 a	595.00 ± 25.98 b	
High dose FeO	57.50 ± 4.78 c	160.00 ± 11.54 c	310.00 ± 39.79 b	0.00 ± 0.00 c	
LSD value	22.194 *	39.613 *	29.850 *	72.893 *	
* (P<0.05).					

As noticed from the results, there's an obvious effect on the hormones levels, while the histological and morphometric study show no significant differences. Possibly, these disparities may due to differences in dosage, chemicals, particle size, and other factors related to the experimental design and the physio-chemical characteristics of the NPs investigated.

The issue concerning alterations in sex hormone levels caused by exposure to some NPs, has also been addressed by some *in vivo* studies which have provided conflicting results. In fact, serum levels of estradiol significantly increased, while progesterone, FSH, LH and testosterone levels diminished in female mice sub-chronically treated with TiO2-NPs [40].

Nevertheless, the administration of zinc oxide NPs to female rats at a concentration of 333.33 mg/kg did not alter serum levels of FSH, LH and estradiol hormones [20]. Some nanoparticles like TiO2-NP induce changes in hormone levels (increased estradiol and decreased progesterone, FSH, LH and testosterone) and cause ovarian injury in female mice may be related to alterations in ovarian gene expression, e.g., Akr1c18, StAr, Cyp17a1, and Lgmn genes were up-regulated. [41]. Little is known about the morphometric alterations induced by nanoparticles, a point to be clarified in the present study.

These results support an idea that iron could play a role in regulation of ovarian function, hormone release, proliferation and apoptosis. [42]. Iron is necessary in many cellular functions, many body processes. in Continuously, the iron accumulates in many tissues such as liver, kidneys [43]. Spleen [44], and uterus [45]. In excess, however, iron is toxic to cells. Free iron ions are extremely toxic and capable of catalyzing many deleterious reactions in cells and tissues [46]. Excess of iron could affect a wide range of mechanisms involved in endometriosis development [47], such as oxidative stress and tissue damage [46], or lesion proliferation [47]. Iron delivered to proliferating cells by transferrin which plays a crucial role in the local regulation of ovarian function and it is important factor for the regulation of granulosa cell differentiation [48].

Cyclin B1 is a marker of proliferation. The expression of cyclin B1 in porcine ovarian granulosa cells is influenced by some metals such as lead, cobalt and iron [46].

Metal-based NPs were reported in many studies to induce changes in reproductive organs histology of laboratory animals and in consequent cause a disruption in reproductive cells production and hormones [49-50]. After 1–5 h of treatment, Au-NPs cause a greater output of estradiol from ovarian granulosa cells, while they decreased the estradiol level after 24 h. TiO2-NP cause changes in hormone levels and progesterone levels in rats.

In addition to the antioxidants released as a response to the oxidative process, nanoparticles may interact with metal-sequestering proteins and antioxidants (from body fluids and intracellularly), that will likely modify the surface properties of the nanoparticle to some extent, rendering them less toxic [51].

Evaluation of these results should therefore take these variables into account. Moreover, when analyzing the endocrine effects on the female reproductive system, it should be noted that the results varies depending on the size of NPs, duration and concentration of the doses.

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ملاجظات :

اختيرت كمية الجرعات المعطاة للفئران اعتمادا على بحوث سابقة وليس اجتهادا وفيها استخدمت وحدة الملغرام لجزيئات اوكسيد الزنك النانوية ووحدة المايكرو غرم لجزيئات اوكسيد الحديد النانوية.

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Molar mass of Fe2O3 = 159.6882 g/mol Molar m, ass of ZnO = 81.3794 g/mol